

## REFERENCES

1. Bohlmann, F., Zdero, C., Bohlmann, R., King, R. M. and Robinson, H. (1980) *Phytochemistry* **19**, 579.
2. Suchy, M. (1962) *Coll. Czech. Chem. Commun.* **27**, 2925.
3. Bohlmann, F., Grenz, M. and Zdero, C. (1977) *Phytochemistry* **16**, 285.
4. Robinson, H. (1983) *Smithsonian Contributions to Botany* **54**, 1.
5. Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1984) *Phytochemistry* **23** (in press).

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## FURTHER EUDESMANOLIDES FROM *DIMEROSTEMMA* SPECIES

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**Key Word Index**—*Dimerostemma asperatum*; *D. bishopii*; Compositae; sesquiterpene lactones; eudesmanolides; degraded syringenin.

**Abstract**—The aerial parts of a new *Dimerostemma* species afforded four eudesmanolides which are all derivatives of arbusculin B and an aldehyde, obviously a degradation product of syringenin. A reinvestigation of *Dimerostemma asperatum* gave a further dimerostemmolide.

From the small Brazilian genus *Dimerostemma* (Compositae, tribe Heliantheae) placed in the subtribe Ecliptinae [1], so far three species have been investigated chemically [2–4]. In addition to more widespread compounds, eudesmanolides with a special substitution pattern were isolated.

The aerial parts of *Dimerostemma bishopii* K. et R. afforded in addition to known compounds minute amounts of the aldehyde **6** and four eudesmanolides (1–4), all being derivatives of arbusculin B. The structure of **6** followed from the  $^1\text{H}$  NMR spectrum (see Experimental). The presence of a symmetrical tetrasubstituted aromatic aldehyde was indicated by the two aromatic protons and the two methoxy groups which displayed sharp singlets. The nature of the oxygen function at C-4 easily could be deduced from the characteristic  $^1\text{H}$  NMR signals. Though no molecular ion was observed in the mass spectrum the structure could be assigned clearly.

The  $^1\text{H}$  NMR spectral data of 1–4 (Table 1) showed that these compounds only differed in the nature of the ester group at C-1 and the substitution at C-15. The signals in the spectrum of the main constituent **2** could be assigned by spin decoupling. Since the broadened double triplet at  $\delta 4.59$  was partly decoupled by irradiation of the signals at  $\delta 2.37$  and  $2.26$  respectively the whole sequence could be established. The nature of the ester groups also followed from the typical  $^1\text{H}$  NMR signals. The relative position of the acetate group in the lactone **3** could be deduced from the mass spectrum which showed elimination of 4-hydroxymethacrylic acid and acetic acid. A 4-*O*-acetate of **2** was absent. Also the  $^1\text{H}$  NMR signal

agreed much better with structure **3**. Accordingly, spin decoupling by irradiation of the H-6 signal showed allylic coupling with down field shifted H-15 signals.

A reinvestigation of the aerial parts of *D. asperatum* Blacke afforded in addition to the compounds isolated previously [3], a further dimerostemmolide (**5**). The  $^1\text{H}$  NMR spectral data (Table 1) showed that the 8-*O*-isobutyrate of the known 1-*O*-[5-hydroxyangelate] of dimerostemmolide was present. The relative position of the ester group was deduced from the chemical shift of H-1 which was identical with that of corresponding lactones with an unsaturated ester group at C-1.

### EXPERIMENTAL

The air dried aerial parts (150 g) of *Dimerostemma bishopii* (voucher RMK 8803, collected in the province Bahia, Brazil) was extracted with  $\text{Et}_2\text{O}$ -petrol, 1:2, and worked-up in the usual fashion [5]. The CC fractions (100 ml) of the extract were as follows: 1 (petrol), 2 ( $\text{Et}_2\text{O}$ -petrol, 1:4), 3 ( $\text{Et}_2\text{O}$ -petrol, 1:1) and 4 ( $\text{Et}_2\text{O}$  and  $\text{Et}_2\text{O}$ -MeOH, 10:1). TLC ( $\text{SiO}_2$ , PF 254, detection by UV light) of fraction 1 and 3 gave only widespread compounds. TLC ( $\text{Et}_2\text{O}$ -petrol, 1:4) of fraction 2 afforded 0.5 mg **6** ( $R_f$  0.52), while repeated TLC of fraction 4 ( $\text{Et}_2\text{O}$ -petrol- $\text{Me}_2\text{CO}$ , 17:2:1) gave 30 mg **2** ( $R_f$  0.52), 24 mg **3** ( $R_f$  0.67) and a mixture of **1** and **4** ( $R_f$  0.72) which after TLC ( $\text{CH}_2\text{Cl}_2$ - $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$ , 1:1:1) gave 4 mg **1** ( $R_f$  0.60) and 0.5 mg **4** ( $R_f$  0.55) (quantities calculated from the  $^1\text{H}$  NMR spectrum of the original mixture of 1–4, separation caused considerable losses).

Table 1.  $^1\text{H}$  NMR spectral data of compounds 1–5 (400 MHz,  $\text{CDCl}_3$ , TMS as int. standard)

	1	2	3	4	5*
H-1	4.89 <i>dd</i>	4.91 <i>dd</i>	4.9 <i>t</i>	4.86 <i>dd</i>	4.95 <i>br dd</i>
H-2	1.95 <i>m</i>	1.95 <i>m</i>	1.95 <i>m</i>	1.92 <i>m</i>	
H-3 $\alpha$	2.34 <i>m</i>	2.37 <i>br dd</i>	2.25 <i>br dd</i>	2.33 <i>m</i>	
H-3 $\beta$		2.26 <i>br d</i>	2.16 <i>br d</i>		
H-6	4.60 <i>br dt</i>	4.59 <i>br dt</i>	4.58 <i>br dt</i>	4.60 <i>br dt</i>	3.89 <i>dd</i>
H-7	2.60 <i>dddd</i>	2.60 <i>dddd</i>	2.64 <i>dddd</i>	2.59 <i>dddd</i>	2.76 <i>dddd</i>
H-8 $\alpha$	2.17 <i>br d</i>	2.15 <i>br d</i>	2.20 <i>br d</i>	2.21 <i>br d</i>	5.21 <i>ddd</i>
H-8 $\beta$	1.70 <i>dddd</i>	1.76 <i>dddd</i>	1.67 <i>dddd</i>	1.70 <i>m</i>	
H-9 $\alpha$	1.83 <i>ddd</i>	1.83 <i>ddd</i>	1.84 <i>ddd</i>	1.85 <i>m</i>	1.67 <i>br dd</i>
H-9 $\beta$	1.51 <i>ddd</i>	1.51 <i>ddd</i>	1.53 <i>ddd</i>	1.54 <i>m</i>	1.82 <i>dd</i>
H-13	6.19 <i>d</i>	6.19 <i>d</i>	6.18 <i>d</i>	6.21 <i>d</i>	6.11 <i>d</i>
H-13'	5.53 <i>d</i>	5.55 <i>d</i>	5.50 <i>d</i>	5.55 <i>d</i>	5.70 <i>d</i>
H-14	1.26 <i>s</i>	1.27 <i>s</i>	1.25 <i>s</i>	1.25 <i>s</i>	1.26 <i>s</i>
H-15	4.45 <i>br d</i>	4.37 <i>br d</i>	5.09 <i>br d</i>	4.46 <i>br d</i>	3.42 <i>d</i>
H-15'	4.17 <i>br d</i>	4.24 <i>br d</i>	4.84 <i>br d</i>	4.17 <i>br d</i>	2.44 <i>d</i>
OCOR	6.08 <i>dq</i>	6.24 <i>br s</i>	6.24 <i>br s</i>	3.10 <i>d</i>	2.41 <i>qq</i>
	5.59 <i>dq</i>	5.84 <i>dq</i>	5.86 <i>dq</i>	2.78 <i>d</i>	1.22 <i>d</i>
	1.95 <i>dd</i>	4.32 <i>br s</i>	4.32 <i>br s</i>	1.59 <i>s</i>	1.13 <i>d</i>
OAc	—	—	2.04 <i>s</i>	— OCOR	6.40 <i>br q</i> , 2.11 <i>br d</i> 4.32 <i>br d</i>

\*H-5, 2.59 *d*.

*J* (Hz): 1, 2 $\alpha$  = 2.5; 1, 2 $\beta$  = 4; 3, 6 = 2; 6, 7 = 11; 7, 8 $\alpha$  = 3; 7, 8 $\beta$  = 12; 7, 13 = 3.2; 7, 13' = 2.6; 8 $\alpha$ , 8 $\beta$  = 13; 8 $\alpha$ , 9 $\alpha$  = 4; 8 $\alpha$ , 9 $\beta$  = 3; 8 $\beta$ , 9 $\alpha$  = 12; 8 $\beta$ , 9 $\beta$  = 3; 9 $\alpha$ , 9 $\beta$  = 13; 15, 15' = 11.5; OMeacr: 3', 3' = 3', 4' ~ 1; OCOC(CH<sub>2</sub>OH)=CH<sub>2</sub>: 3', 3' = 3', 4' ~ 1; OEpoxyiBu: 3', 3' = 6; compound 5: 1, 2 = 2.5; 5, 6 = 6, 7 = 7, 8 = 11; 7, 13 = 3.5; 7, 13' = 3; 8, 9 $\alpha$  = 11; 8, 9 $\beta$  = 4; 15, 15' = 4.3; OiBu: 2, 3 = 2, 4 = 7; 5-hydroxy ang: 3, 4 = 7; 5, OH ~ 5.

A reinvestigation of the polar fractions of the aerial parts (170 g) of *Dimerostemma asperatum* (voucher RMK 8951) gave in addition to dimeroperatic acid and the eudesmanolides isolated previously [3] by HPLC (RP 8, MeOH–H<sub>2</sub>O, 7:3) of the TLC fraction containing the eudesmanolides [3] (Et<sub>2</sub>O) 1 mg of the dimerostemmolide 5 (*R*<sub>f</sub> 3.5 min). Known compounds were identified by comparing the  $^1\text{H}$  NMR spectra with those of authentic material. Due to the very small amounts compounds 1–6 could not be induced to crystallize.

15-Hydroxy-1 $\alpha$ -methacryloyloxy-arbusculin B (1). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3580 (OH), 1765 ( $\gamma$ -lactone), 1705 (C=CCO<sub>2</sub>R); MS *m/z* (rel. int.): 246.126 [M – RCO<sub>2</sub>H]<sup>+</sup> (17) (C<sub>15</sub>H<sub>18</sub>O<sub>3</sub>), 228 [246 – H<sub>2</sub>O]<sup>+</sup> (48), 213 [228 – Me]<sup>+</sup> (20), 69 [C<sub>3</sub>H<sub>5</sub>CO]<sup>+</sup> (100); CI (isobutane): 333 [M + 1]<sup>+</sup> (28), 315 (333 – H<sub>2</sub>O)<sup>+</sup> (78), 229 [315 – RCO<sub>2</sub>H]<sup>+</sup> (100).

15-Hydroxy-1 $\alpha$ -[4-hydroxymethacryloyloxy]-arbusculin B (2). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3595 (OH), 1770 ( $\gamma$ -lactone), 1710 (C=CCO<sub>2</sub>R); MS *m/z* (rel. int.): 348.157 [M]<sup>+</sup> (0.3) (C<sub>19</sub>H<sub>24</sub>O<sub>6</sub>), 330 [M – H<sub>2</sub>O]<sup>+</sup> (4), 246 [M – RCO<sub>2</sub>H]<sup>+</sup> (38), 228 [330 – RCO<sub>2</sub>H]<sup>+</sup> (100), 213 [228 – Me]<sup>+</sup> (32), 85 [RCO]<sup>+</sup> (82).

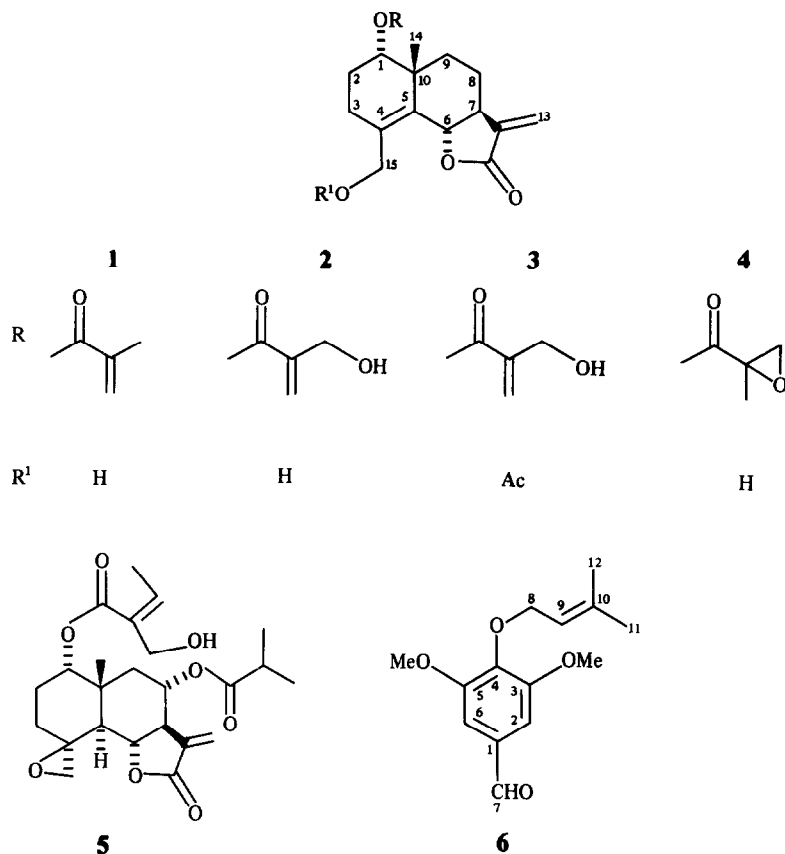
15-Acetoxy-1 $\alpha$ -[4-hydroxymethacryloyloxy]-arbusculin B (3). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3600 (OH), 1775 ( $\gamma$ -lactone), 1730 (OAc), 1710

(C=CCO<sub>2</sub>R); MS *m/z* (rel. int.): 330.147 [M – HOAc]<sup>+</sup> (3.5) (C<sub>19</sub>H<sub>22</sub>O<sub>5</sub>), 288 [M – RCO<sub>2</sub>H]<sup>+</sup> (2), 246 [288 – ketene]<sup>+</sup> (22), 228 [288 – HOAc]<sup>+</sup> (100), 213 [228 – Me]<sup>+</sup> (27), 85 [RCO]<sup>+</sup> (37); CI (isobutane): 391 [M + 1]<sup>+</sup> (3), 331 [391 – HOAc]<sup>+</sup> (100), 229 [331 – RCO<sub>2</sub>H]<sup>+</sup> (34); [ $\alpha$ ]<sub>D</sub><sup>20</sup> = +16 (c 1.0, CHCl<sub>3</sub>).

15-Hydroxy-1 $\alpha$ -[2,3-epoxyisobutyryloxy]-arbusculin B (4). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3600 (OH), 1770 ( $\gamma$ -lactone), 1730 (CO<sub>2</sub>R); MS *m/z* (rel. int.): 246.126 [M – RCO<sub>2</sub>H]<sup>+</sup> (4) (C<sub>15</sub>H<sub>18</sub>O<sub>3</sub>), 228 [246 – H<sub>2</sub>O]<sup>+</sup> (10), 85 [RCO]<sup>+</sup> (38), 57 [85 – CO]<sup>+</sup> (100); CI (isobutane): 349 [M + 1]<sup>+</sup> (7), 331 [349 – H<sub>2</sub>O]<sup>+</sup> (100).

8-O-Isobutyryl-dimerostemmolide-1-O-[5-hydroxyangelate] (5). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3595 (OH), 1770 ( $\gamma$ -lactone), 1720 (C=CCO<sub>2</sub>R); MS *m/z* (rel. int.): 378 [M – O=C=C(Me)<sub>2</sub>]<sup>+</sup> (1), 360, 157 [M – C<sub>3</sub>H<sub>7</sub>CO<sub>2</sub>H]<sup>+</sup> (3) (C<sub>20</sub>H<sub>24</sub>O<sub>6</sub>), 332 [M – C<sub>4</sub>H<sub>6</sub>(OH)CO<sub>2</sub>H]<sup>+</sup> (1.5), 244 [332 – C<sub>3</sub>H<sub>7</sub>CO<sub>2</sub>H]<sup>+</sup> (32), 99 [C<sub>4</sub>H<sub>6</sub>(OH)CO]<sup>+</sup> (100), 71 [C<sub>3</sub>H<sub>7</sub>CO]<sup>+</sup> (18).

3,5-Dimethoxy-4-hydroxybenzaldehyde 3',3'-dimethylallyl ether (6). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 2720, 1685 (CHO); MS *m/z* (rel. int.): 182 [M – isoprene]<sup>+</sup> (66), 69 [Me<sub>2</sub>C=CHCH<sub>2</sub>]<sup>+</sup> (100);  $^1\text{H}$  NMR (CDCl<sub>3</sub>): 9.87 *s* (CHO), 5.55 *br t* (H-9, *J* = 7 Hz), 4.68 *br d* (H-8, *J* = 7 Hz), 1.68 *br s* and 1.59 *br s* (H-11, H-12), 3.93 *s* (6H, OMe), 7.12 *s* (2H, H-2, H-6).



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#### REFERENCES

1. Robinson, H. (1981) *Smithsonian Contributions to Botany* **51**, 1.
2. Bohlmann, F., Dhar, A. K., Jakupovic, J., King, R. M. and Robinson, H. (1981) *Phytochemistry* **20**, 838.
3. Bohlmann, F., Ziesche, J., King, R. M. and Robinson, H. (1981) *Phytochemistry* **20**, 1335.
4. Bohlmann, F., Singh, P., Jakupovic, J., King, R. M. and Robinson, H. (1982) *Phytochemistry* **21**, 1343.
5. Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1984) *Phytochemistry* **23** (in press).